

2.1.a

Yoctomole = $6.022E(23-24) = .6$ atoms

2.1.b

Nanocentury to seconds = $3.154E7(+2 \text{ (century)} -9 \text{ (nano)}) = 3.154E0$ is almost pi

2.2

A CD is half a gigabyte and a millimeter thick. A petabyte is $2E6$ CDs, 2km tall. This is more than twice as tall as our tallest buildings.

2.3

$2^{(10^{80})}$ atoms in universe) = $(2^{10})^{80} = 1E(3^{*}80) = 1E240$

Or $2^{(10^{80})} = 10^{(80/\log_{10}(2))} = 10^{265}$.

2.4

$6.673E-11 \text{ (m}^3/\text{kg s}^2) * 1 \text{ kg} / (1\text{m})^2 = 6.6E-11 \text{ m/s}^2 \text{ vs } 9.8\text{m/s}^2$

$6.8e-12 \rightarrow 10^{(-12 + \log(6.8))} \text{ dB} = (-120 + 10^{*.83})\text{dB} = -111.7 \text{ dB}$

2.5a

Assume TNT is pure Nitrogen with 3 bonds, each storing 1eV, shared with another atom.

$(907,000 \text{ g/Ton}) / (14 \text{ grams per mole}) * (6.022e23 \text{ atoms per mole}) * (1.5\text{eV per atom}) * (1.6E-19 \text{ J/eV})$

+6 -1.1 +23.8 +.1 -18.8

$10^{(10)}$ Joules per Ton tnt

Overestimated by roughly 10x.

2.5b

10,000 tons of TNT is $10^{(10+4)}$ Joules

Assume Uranium is 10^6 eV per atom, and 237 grams per enriched mole.

$(10^{(14)} \text{ Joules}) / (1.6E-19 \text{ J/eV}) / (1E6 \text{ eV per atom}) / (6.022e23 \text{ atoms per mole}) * (237 \text{ g per mole})$

+14 +18.8 -6 -23.8 +2.3

$10^{(5.3)}$ grams of Uranium, or 200kg. Correcting for my TNT mis-estimation brings its to $10^{(4.3)}$ grams, or 20kg

Little Boy was 4,400 kg and yielded 15 kilotons, but only contained 64 kg uranium. This is 4.3 kilotons per kilogram, compared to my half kiloton per kilogram.

2.5c.

$1\text{kg} * (3E8 \text{ m/s})^2 = 9E16 \text{ Joules per kilogram for total conversion.}$

$1E13 \text{ Joules} / 20\text{kg} = 5E11 \text{ Joules per kilogram as a nuclear bomb. Only one part in 180,000 is used.}$

2.6

$h = 6.26E-34 \text{ Joule Seconds} \rightarrow -33.2$

Assume baseball is 100 grams and 50m/s

$6.26E-34 \text{ Js} / (.100 \text{ kg} * 50\text{m/s})$

$-33.2 \quad +1 \quad -1.7$

$10^{(-33.9)} \text{ meters} = 1.25E-34 \text{ meters}$

2.6b

Each degree of freedom of a molecule has on average $3/2 \text{ kT}$ kinetic energy, and weighs 28AMU

$3/2 * (300 \text{ Kelvin}) * (1.38E-23 \text{ J/K}) = 6.2E-21 \text{ Joules per molecule}$

$V = \text{sqrt}(2 * E / m) = \text{sqrt}(2 * 6.2E-21 \text{ J} / (28 \text{ AMU} * 1.66E-24 \text{ grams per AMU})) = 520\text{m/s}$

2.6c

$PV = nRT$

At STP, 1 mole is 22.4L.

$1 \text{ mole} * 6.022E23 / 22.4\text{L} = 2.69E25 \text{ per m}^3.$

$$\sqrt[3]{\frac{\text{m}^3}{2.69E25}} \approx 3\text{nm}$$

2.6d

Distance remains constant at 3nm with falling temperature

$$\lambda = h/p$$

$$p = \sqrt{2mKE} = \sqrt{2m\left(\frac{3}{2}kT\right)}$$

$$\sqrt{3mkT} = h/\lambda$$

$$T = \left(\frac{h}{\lambda}\right)^2 \frac{1}{3mk} = \left(\frac{6.626\text{Js}}{3\text{nm}}\right)^2 \frac{1}{3(28*1.66E-24\text{g})(1.38E-23\text{J/K})} = .025\text{ Kelvin}$$

2.7a

$$\frac{1}{2}mv^2 = GMm/r$$

$$v = \sqrt{2GM/r}$$

2.7b

$$r = \frac{2GM}{v^2} \rightarrow \frac{2GM}{c^2}$$

2.7c

$$\lambda = \frac{hc}{E} = \frac{hc}{Mc^2} = h/Mc$$

2.7d

$$\lambda = r$$

$$\frac{h}{Mc} = \frac{2GM}{c^2}$$

$$h = \frac{2GM^2}{c}$$

$$M = \sqrt{hc/2G} = 38.6\mu\text{g}$$

2.7e

$$\frac{2GM}{c^2} = 38.6\mu\text{g} * \frac{2G}{c^2} = 5.73E-35\text{meters}$$

2.7f

$$E = \frac{hc}{\lambda} = h \frac{c^3}{2GM} = 3.46E9\text{J}$$

2.7g

$$\text{Period} = \frac{\lambda}{c} = 1.79E-43\text{ seconds}$$

2.8A

Radius of Sphere is $L/2$

Construct triangle through edges (45 degrees off vertical).

Base is $\sqrt{2}L$

Dashed line is sphere radius $\frac{L}{2}$

Triangles are similar (see labeling)

$$A = \sqrt{B^2 - C^2} \quad A' = \sqrt{1/2}L$$

$$B = A' = \sqrt{1/2}L \quad B' =$$

$$C = \frac{1}{2}L \quad C' = H$$

$$A = L\sqrt{1/2 - 1/4} = L/2$$

$$\frac{c'}{c} = \frac{A'}{A}$$

$$C' = \frac{(L/\sqrt{2})(L/2)}{(L/2)} = L/\sqrt{2} = H$$

$$B' = \frac{A'}{A}B = \frac{(L/\sqrt{2})(L/\sqrt{2})}{(L/2)} = L$$

2.8B

Construct as hemisphere missing 4 partial spheres

$\left(\frac{1}{2}\right)\left(\frac{4}{3}\right)\pi r^3 - 4\left(\frac{\pi h}{6}\right)(3r'^2 + h^2)$, r = big sphere radius, r' = radius of base of partial sphere, h = extension above face

r is known to be $L/2$

The face is an equilateral triangle, L on each side (see calculation of B' above).

$$r' = \frac{1}{2}L \tan^{-1} 60 = L/\sqrt{12}$$

Type equation here.

The height of the sphere cap is the major radius minus the distance to the face d

$$d = L\sqrt{\frac{1^2}{2} - \frac{1}{12}} = L/\sqrt{6}$$

$$h = r - d = L/2 - L/\sqrt{6}$$

All together

$$\left(\frac{1}{2}\right)\left(\frac{4}{3}\right)\pi r^3 - 4\left(\frac{\pi h}{6}\right)(3r^2 + h^2)$$

$$\left(\frac{1}{2}\right)\left(\frac{4}{3}\right)\pi(L/2)^3 - 4\left(\frac{\pi(L/2 - L/\sqrt{6})}{6}\right)\left(3\left(\frac{L}{\sqrt{12}}\right)^2 + (L/2 - L/\sqrt{6})^2\right)$$

$$\approx 0.2124L^3$$